

## Experiments Directed toward a Total Synthesis of Dynemicin A: A Solution to the Stereochemical Problem

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**Summary:** The stereochemical issues required for a total synthesis of dynemicin A (1) have been resolved by employing an intramolecular Diels–Alder reaction and a diastereoselective addition of a magnesioacetylide to a quinoline under the influence of methyl chloroformate.

Given the fascinating structure, the potent cytotoxicity, and the novel bioorganic chemistry of dynemicin A (1), the interest elicited by this metabolite derivative of *Micromonospora chersina* is not surprising.<sup>2,3</sup> Much of this research has been directed toward the conception, synthesis, and evaluation of structures which, while less complex than dynemicin, still partake of its DNA cleavage capacity.<sup>4</sup> It is presumed<sup>5</sup> that this cutting action involves reduction of the anthraquinone, epoxide solvolysis, enediyne → diyl rearrangement,<sup>6</sup> and intermolecular hydrogen atom transfer, from oligodeoxyribonucleotide targets (2 → 4). Impressive strides in dynemicin-inspired biomimetic chemistry have been registered by Nicolaou<sup>4,7</sup> and Wender.<sup>8</sup>

Another facet of the dynemicin problem which has prompted considerable effort has been the challenge of total synthesis. A variety of interesting approaches have been devised in response to this most worthy undertaking.<sup>9</sup> The ranking accomplishment toward this end is the

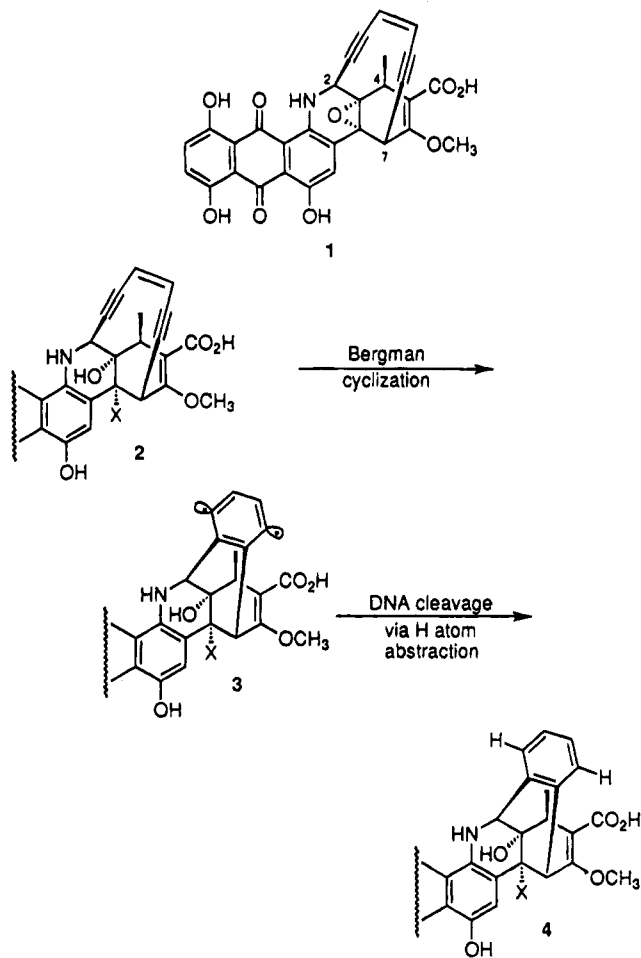


Figure 1.

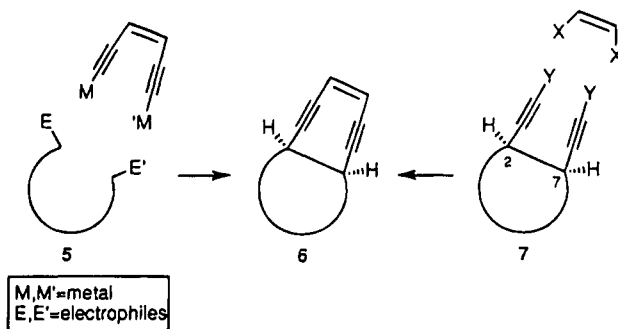


Figure 2.

remarkable total synthesis of various O-methylated derivatives of dynemicin methyl ester accomplished by Schreiber and associates.<sup>10</sup> In this paper, we describe a unique solution to the stereochemical issues which must

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(1) Recipient of a predoctoral fellowship from Memorial Sloan-Kettering Cancer Center.

(2) Konishi, M.; Ohkuma, H.; Matsumoto, K.; Tsuno, T.; Kamei, H.; Miyaki, T.; Oki, T.; Kawaguchi, H.; VanDuyne, G. D.; Clardy, J. *J. Antibiot.* **1989**, *42*, 1449. Konishi, M.; Ohkuma, H.; Tsuno, T.; Oki, T.; VanDuyne, G. D.; Clardy, J. *J. Am. Chem. Soc.* **1990**, *112*, 3715.

(3) For an excellent review of the chemistry and biology associated with the enediyne antibiotics, see: Nicolaou, K. C.; Dai, W.-M. *Angew. Chem., Int. Ed. Engl.* **1991**, *30*, 1387.

(4) Nicolaou, K. C.; Dai, W.-M.; Tsay, S.-C.; Estevez, V. A.; Wrasildo, W. *Science* **1992**, *256*, 1172.

(5) Semmelhack, M. F.; Gallagher, J.; Cohen, D. *Tetrahedron Lett.* **1990**, *31*, 1521. Sugiura, Y.; Shiraki, T.; Konishi, M.; Oki, T. *Proc. Nat. Acad. Sci. U.S.A.* **1990**, *87*, 3831.

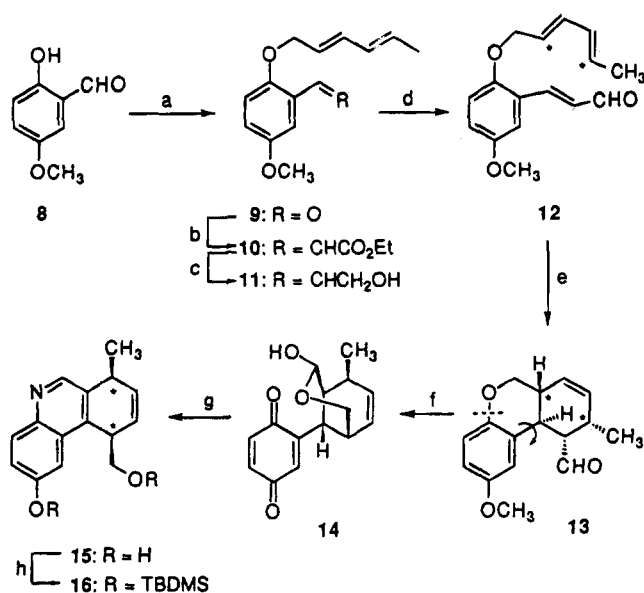
(6) Jones, R. R.; Bergman, R. G. *J. Am. Chem. Soc.* **1972**, *94*, 660. Bergman, R. G. *Acc. Chem. Res.* **1973**, *6*, 25.

(7) Nicolaou, K. C.; Dai, W.-M.; Wendeborn, S. V.; Smith, A. L.; Torisawa, Y.; Maligras, P.; Hwang, C.-K. *Angew. Chem., Int. Ed. Engl.* **1991**, *30*, 1032. Nicolaou, K. C.; Maligras, P.; Suzuki, T.; Wendeborn, S. V.; Dai, W.-M.; Chadha, R. K. *J. Am. Chem. Soc.* **1992**, *114*, 8890. Nicolaou, K. C.; Dai, W.-M. *J. Am. Chem. Soc.* **1992**, *114*, 8908.

(8) Wender, P. A.; Zercher, C. K.; Beckham, S.; Haubold, E.-M. *J. Org. Chem.* **1993**, *58*, 5867.

(9) (a) Porco, J. A., Jr.; Schoenen, F. J.; Stout, T. J.; Clardy, J.; Schreiber, S. L. *J. Am. Chem. Soc.* **1990**, *112*, 7410. (b) Chikashita, H.; Porco, J. A., Jr.; Stout, T. J.; Clardy, J.; Schreiber, S. L. *J. Org. Chem.* **1991**, *56*, 1692. (c) Wood, J. L.; Porco, J. A., Jr.; Taunton, J.; Lee, A. Y.; Clardy, J.; Schreiber, S. L. *J. Am. Chem. Soc.* **1992**, *114*, 5898. (d) Nicolaou, K. C.; Hwang, C.-K.; Smith, A. L.; Hwang, S. V. *J. Am. Chem. Soc.* **1991**, *113*, 3106. (e) Wender, P. A.; Zercher, C. K. *J. Am. Chem. Soc.* **1991**, *113*, 2311. (f) Magnus, P.; Fortt, S. M. *J. Chem. Soc., Chem. Commun.* **1991**, 544. (g) Nishikawa, T.; Isobe, M.; Goto, T. *Synlett* **1991**, 393. (h) Nicolaou, K. C.; Gross, J. L.; Kerr, M. A.; Lemus, R. H.; Ikeda, K.; Ohe, K. *Angew. Chem.* **1994**, *106*, 790.

(10) Taunton, J.; Wood, J. L.; Schreiber, S. L. *J. Am. Chem. Soc.* **1993**, *115*, 10378.

Scheme 1<sup>a</sup>

<sup>a</sup> Key: (a) (*E,E*)-sorbyl bromide, K<sub>2</sub>CO<sub>3</sub>, acetone, reflux, 0.5 h (96%); (b) NaH, (EtO)<sub>2</sub>P(O)CH<sub>2</sub>CO<sub>2</sub>Et, THF, 0 °C to rt, 0.5 h (95%); (c) DIBALH, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, 1 h (94%); (d) (COCl)<sub>2</sub>, DMSO, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C to rt (92%); (e) ZnCl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt, 72 h (60%); (f) CAN, H<sub>2</sub>O, CH<sub>3</sub>CN, 0 °C, 0.5 h (62%); (g) NH<sub>4</sub>OAc, AcOH, H<sub>2</sub>O, 100 °C, 0.5 h; 0 °C, NH<sub>4</sub>OH; (h) TBDMSCl, imidazole, CH<sub>2</sub>Cl<sub>2</sub>, DMF, 0 °C to rt, 2 h (86%) over two steps.

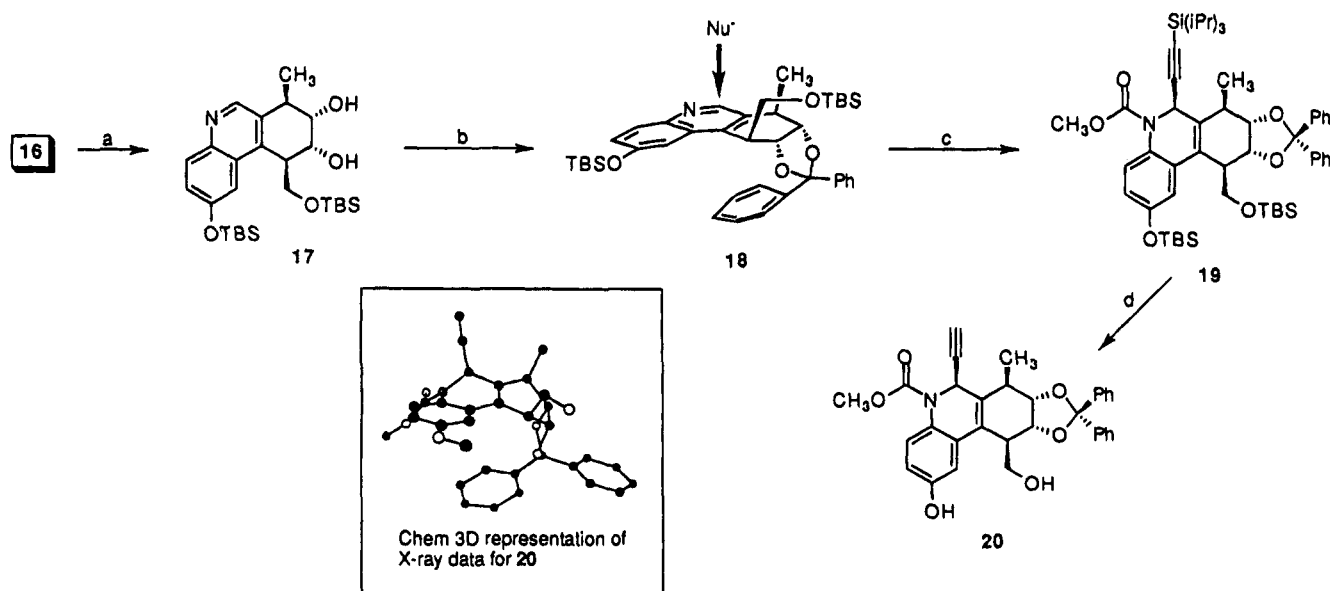
be addressed in any total synthesis undertaking directed toward dynemicin.

In our total synthesis of calicheamicinone<sup>11</sup> we had introduced the entire enediyne unit through exploitation of the bis nucleophilic potential of a *cis*-diethynylethylene moiety (see Figure 2) (5 → 6).<sup>11,12</sup> Happily, other workers in the dynemicin area have applied this central paradigm in one form or another toward the assembly of various analog constructs.<sup>9d,e,g</sup> Of course, if such a scheme could be realized, the required *cis* configurational relationship between C<sub>2</sub> and C<sub>7</sub> (dynemicin numbering) is, perforce, solved. While we have directed some effort toward

application of this procedure to the dynemicin area,<sup>13</sup> our focus of late has centered on a conceptually different idea. We wondered whether *cis*-related appendages at C<sub>2</sub> and C<sub>7</sub> could be mobilized for cyclization to produce the enediyne. Implementation of such a plan (see Figure 2) (7 → 6) is contingent on prearrangement of the *cis* relationship of these rather remote loci. Moreover, these stereogenic centers must be coordinated with the C<sub>4</sub> site (dynemicin numbering) bearing the secondary methyl group. We set as an interim goal the synthesis of compound 19 (Scheme 2) in which these questions are addressed.

Alkylation of 5-methoxysalicylaldehyde (8)<sup>14</sup> with (*E,E*)-sorbyl bromide<sup>15</sup> afforded a 96% yield of 9. Emmons condensation of aldehyde 9 with ethyl diethylphosphonoacetate set the stage for the synthesis of enal 12 as shown. The action of the latter with zinc chloride in methylene chloride afforded a 59% yield of 13 with only trace quantities of *exo* isomer.<sup>16</sup> The suprafaciality of the Diels–Alder reaction was thus used to produce the *cis* C<sub>4</sub>–C<sub>7</sub> connectivity for dynemicin as encoded in the appropriate centers in 13. The relationship is more readily recognized in 15 which was obtained in two steps from 13. In the first step, CAN oxidation of 13 leads to 14. Reaction of the latter with ammonium acetate produced 15 which was then directly converted to the corresponding TBS ether 16 in 86% overall yield.

We next dealt with installation of the required stereochemistry at C<sub>2</sub>. Toward this end, compound 16 was treated with catalytic osmium tetroxide,<sup>17</sup> thereby affording 17 as the only isolated diol. The vicinal hydroxyl groups were engaged as a benzhydrylidene acetal (see 18). We hoped that the  $\alpha$ -face of the molecule would now be sufficiently hindered so as to encourage ethynylation of the azomethine linkage from the  $\beta$ -face. In the event, treatment of 18 with bromomagnesium (triisopropylsilyl)acetylide and methyl chloroformate<sup>18</sup> afforded 19, whose structure was proven by crystallographic analysis of its tris-desilylated product 20. In addition to confirming the configurations of 20 as well as showing that the chirality has been properly arranged

Scheme 2<sup>a</sup>

<sup>a</sup> Key: (a) OsO<sub>4</sub>, NMO, THF, *t*-BuOH, 4 h, rt (90%); (b) Ph<sub>2</sub>(OMe)<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 1.1 equiv of H<sub>2</sub>SO<sub>4</sub>, 40 °C, 2 h (90%); (c) (triisopropylsilyl)acetylene, EtMgBr, THF, rt, 2 h; then -78 °C, add 18, ClCO<sub>2</sub>CH<sub>3</sub>, -20 °C, 20 h (75%  $\beta$ -diastereomer; 12%  $\alpha$ -diastereomer).

for reaching dynemicin A, the crystal structure suggests why the nucleophilic addition of the magnesio (triiso-

(11) Haseltine, J. N.; Cabal, M. P.; Mantlo, N. B.; Iwasawa, N.; Yamashita, D. S.; Coleman, R. S.; Danishefsky, S. J.; Schulte, G. K. *J. Am. Chem. Soc.* **1991**, *113*, 3850.

(12) Danishefsky, S. J.; Yamashita, D. S.; Mantlo, N. B. *Tetrahedron Lett.* **1988**, *29*, 4681.

(13) Yoon, T. Ph.D. Thesis, Yale University, 1994.

(14) Available from Lancaster Synthesis, Windham, NH 03087.

(15) Mori, K. *Tetrahedron* **1974**, *30*, 3807.

(16) A higher yield in the Diels–Alder cycloaddition (93%) was obtained when the reaction was performed in the absence of catalyst by refluxing in benzene. However, an erosion in the endo selectivity (3:1 endo:exo) was observed under the thermal conditions. Although, in principle, the required C<sub>4</sub>–C<sub>7</sub> relationship also pertains in the exo product, in practice this isomer does not survive the subsequent CAN oxidation presumably due to its inability to form a hemiacetal.

(17) VanRheenan, V.; Kelly, R. C.; Cha, D. Y. *Tetrahedron Lett.* **1976**, *17*, 1973.

(18) Yamaguchi, R.; Nakazono, Y.; Kawanisi, M. *Tetrahedron Lett.* **1983**, *24*, 1801.

propylsilyl)acetylide had occurred from the  $\beta$ -face. The conversion of **19** to a des DE congener of dynemicin A is described in the following paper in this issue.

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**Supplementary Material Available:** <sup>1</sup>H NMR spectra, IR, and MS data are available for compounds **8**–**19** (13 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead for ordering information.